Investigation of Copper Binding to Hydrosoluble Proteins of an Amphipod

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ABSTRACT

This study investigated the body copper accumulation in the amphipod Gammarus sp.. Groups of copper-binding proteins were also investigated considering their molecular weights. Regarding copper accumulation, a saturation type kinetics was observed with $K_m$ and $V_{max}$ values of 0.48 mg Cu.L$^{-1}$ and 713.3 µg Cu.g$^{-1}$ dw, respectively. Size exclusion chromatography (5 to 150 KDa) associated with ICP-MS showed that most of copper-binding proteins in Gammarus sp. have molecular weights <12 KDa, probably corresponding to metallothioneins.

Keywords: copper accumulation, Gammarus sp., ICPMS, metallothioneins, size exclusion chromatography.

INTRODUCTION

Aquatic invertebrates can accumulate metals in their body, at different concentrations, whether or not this metal is essential to the metabolism. In the case of crustaceans, tissue and body concentrations of trace metals vary greatly (Rainbow et al., 1989). Specifically to amphipods, both essential and non-essential metals are accumulated without excretion (Rainbow et al., 1989).

Trace metals potentially bind to any macromolecule exhibiting affinity for them. Since they have typically affinity for sulphur and nitrogen (Nieboer & Richardson, 1980), and proteins are made up of amino acids containing these chemicals, there is no shortage of potential binding sites for trace metals in cells (Rainbow, 1997). Metal detoxification often involves binding proteins, such as metallothioneins (Viarengo, 1989). These proteins have a low molecular weight and they are capable to sequester metals, primarily copper, zinc, cadmium and mercury. Functions of metallothioneins are not completely understood, but it is accepted that their main function is...
associated with the homeostase of essential metals, such as copper and zinc (Brady, 1982).

**MATERIAL AND METHODS**

Gammarid amphipods (*Gammarus sp.*, 125.5 ± 20.1 mg) were collected in the Saint Lawrence Estuary (Rimouski, Quebec, Canada), and adult males were transported to the laboratory for acclimation in filtered sea water (natural water from Saint Laurance, 0.7-1.0 µm), at salinity 27 and 14 °C of temperature. Copper (CuCl₂.2H₂O) exposure was performed via water for 5 days, in duplicate, using glass flasks containing 500 mL of experimental medium continuously aerated, with 10 amphipods in each flask, in a semi-static condition (medium exchanged each 24 h). Nominal metal concentrations added to the water were 0.1, 0.5, 1.0, 5.0 and 10.0 mg Cu.L⁻¹.

After exposure, the surviving amphipods from each tested concentration were cleaned in salt water without copper, dried with paper, pooled, lyophilized, powdered, weighed and prepared for three different analyses:

a) Total copper accumulation, where an aliquot of whole-body *Gammarus* powder (50 mg) after acid digestion (1 mL of HNO₃ and 0.5 mL of H₂O₂ for 2 h in water bath at 80 °C) by Inductive Coupled Plasma – Mass Spectrometry (ICP-MS);

b) Protein molecular weights (MWs), after aqueous extraction, where sub-samples (0.1 g) from *Gammarus* powder were extracted with 1.4 mL of water by shaking in a water bath at 37 °C for 24 h. After extraction, samples were centrifuged at 10,000 × g for 20 min. Supernatants were analyzed on an HPLC column (5-150 KDa, Alltech Prosphere 125 HR) by Size Exclusion Chromatography (SEC). Injected volume was 100 µL. MWs were determined by relating the time of protein peaks from samples with that from standard proteins used for calibration (Sigma, MW-GF-200: Gel Filtration Molecular Weight Markers 12440 to 200000 Da); and

c) Distribution of copper-binding proteins, determined using the same SEC column coupled to ICP-MS, following the same chromatographic conditions of pressure and elution rate. Retention time of copper peaks (detected by ICP-MS) were related with those from proteins (detected by spectrophotometry), indicating the MWs of copper-binding proteins.

**RESULTS AND DISCUSSION**

The average mortalities after 5 days of experiment in each tested copper concentration were: 0, 20, 45, 60, 50 and 70%, in control (no copper exposure), 0.1, 0.5, 1.0, 5.0 and 10.0 mg Cu L⁻¹, respectively. Different percentage of mortality were tested with intention to evaluate until which concentration organisms have capacity to use defense strategies against copper exposure. It was not possible to kill more than 70% of organism in a soluble copper concentration. Regarding the whole body copper burden in surviving amphipods, a clear saturation type kinetics as a function of the metal concentration in the experimental medium was observed. Kₘ and Vₘₐₓ values were calculated as 0.48 mg Cu.L⁻¹ and 713.3 mg Cu.g⁻¹ dw, respectively. Interpretation of whole body copper accumulation kinetics indicates that there is a simple adsorption of the dissolved metal into the biological tissues, where the metal binds with an affinity (Kₘ) for sites available up to a certain limit of saturation (Vₘₐₓ).

A typical protein chromatogram for control amphipods and retention time of standard proteins is shown in Figure 1a. Apart from the chromatogram corresponding to the highest concentration tested (10 mg Cu.L⁻¹), where a different peak in a retention time between 16 and 18 min was observed (Figure 1b), all chromatograms were very similar to the control one (Figure 1a).

After analysis with ICP-MS, chromatograms showed an increase in the peak representing the copper-binding proteins paralleled by a decrease in the zinc peak at retention time between 12 and 16 min after exposure to 10 mg Cu.L⁻¹ for 5 days (Figure 2).

This Zn-Cu exchange was tested putting copper (equivalent to 10 mg.L⁻¹) directly on a sample of powdered control organism and the metals exchange was not verified. It is an indication that this Zn-Cu exchange was during copper exposure and an internal behavior. Chromatograms for the other copper
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Figure 2 – Chromatogram of Gammarus sp. a) maintained under control conditions; or b) exposed to 10 mg Cu.L⁻¹ for 5 days. Chromatogram was obtained by Size Exclusion Chromatography (SEC) column (Alltech Prosphere 125 HR analytical column) coupled to Inductive Coupled Plasma – Mass Spectrometry (ICP-MS). Peaks were detected by ICP-MS for five different metals (Mn, Fe, Cu, Zn, and As).

concentrations tested are not shown. However, it was observed a gradual increase in the copper peak concomitant with a reduction in the zinc peak, as the copper concentration was raised in the experimental medium. This finding suggests that there is a replacement of zinc by copper in the same range of retention time, i.e., in the same range of protein MWs, as...
copper concentration was raised in the experimental medium. Binding affinity of different metals for metallothioneins varies considerably, but a pattern of increasing affinity is reported as $\text{Zn} < \text{Pb} < \text{Cd} < \text{Cu} < \text{Ag} = \text{Hg} = \text{Bi}$, thereby making zinc readily displacable by other metals, including copper (Kagi & Kojima, 1987).

Comparing chromatograms of the extreme situations tested (control conditions and exposure to 10 mg Cu.L$^{-1}$), it was observed a higher quantity of proteins and peaks representing copper-binding proteins with MWs < 12 KDa. This result is in close agreement with that reported for the Antarctic krill (Li et al., 2005). Using different SEC columns, these authors verified that the highest copper distribution (59%) was associated with proteins of MWs between 2 and 20 KDa, followed (36%) by those with MWs < 2 KDa. Worth noting is that these results were obtained from animals collected directly from the field, without experimental copper exposure in laboratory. This finding clearly indicates that a baseline amount of copper-binding proteins is normally present under natural conditions. Under the experimental conditions employed in the present study, it is likely that the amount of these proteins was increased after experimental exposure to copper, resulting in a higher copper accumulation in tissues than that observed under natural conditions. Since both studies used similar techniques to analyze whole body copper accumulation, when krill data (Li et al., 2005) are compared with data from control amphipods (Gammarus sp.) tested in the present study, an ~10-fold higher copper accumulation is observed in the gammarids (krill = 7.4 and Gammarus sp. = 78.4 µg Cu.g$^{-1}$ dw). Other values reported in the literature for two krill species from Central California (Sydeman & Jarman, 1998; Nygard et al., 2001), being closer to that reported for gammarids in the present study.

Metal detoxification often involves binding proteins, such as metallothioneins (Viarengo, 1989). Authors indicate that metallothioneins play an important role in the homeostasis of essential metals such as Cu and Zn; they can act as essential metal stores to fulfill enzymatic and other metabolic demands as well as in the sequestration of nonessential metals, such as Cd and Hg (Amiard, et al., 2006; Coyle et al., 2002). Copper is known to bind to metallothioneins or metallothioneins-like proteins, which in mammals and other vertebrates usually have a range of MWs from 6-15 KDa. Data reported in the present study showed that in Gammarus sp. copper was mostly bound to water soluble proteins with MWs between 30 and 12 KDa and < 12 KDa. Proteins with MWs < 12 KDa might be equivalent to the copper-binding metallothioneins in Gammarus sp., but, as observed at present results, other proteins are also involved in copper accumulation in this species. This statement is based on the fact that proteins with MWs > 15 KDa (maximum MWs of metallothioneins reported in the literature) were shown to be bound to copper. The change in protein chromatograms observed at the highest copper concentration tested, associated with the copper peaks detected by ICP-MS at the same MWs range, might be an indication of a mobilization of some unknown proteins, other than metallothioneins, to counteract the exposure to high copper concentration.

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